## SHORT COMMUNICATION

## STRUCTURE, DIELECTRIC AND PHOTOELASTIC PROPERTIES OF GLASSES IN THE SYSTEM Ge-Sb-S \*

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Radial atomic distribution curves, transmission spectra, optical absorption edge, dielectric and photoelastic properties of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses are studied. The short - range atomic order is shown to be related to the optical, dielectric and photoelastic parameters of the glasses.

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The possibility to apply the glasses from the system Ge-Sb-S in optical device construction due to their high photosensitivity, transparency in the visible, near- and medium - infrared (IR) range is the reason for the interest to their studies.

In the present paper the results of radial atomic distribution curves (RADC), IR transmission spectra, optical absorption edge, dielectric and properties of quasibinary  $GeS_2-Sb_2S_3$  glasses are reported.

The interest to the studies of the glasses of this particular section of the Ge-Sb-S system is determined first of all by the difference of the short-range order and properties for the pure binary compounds. The structure of glassy  $GeS_2$  is studied rather extensively. The glass matrix is build mostly by  $GeS_4$  tetrahedra linked together by two - fold coordinated sulphur atoms. The first coordination sphere radius  $R_1$ , determined from RADC, for  $GeS_2$  glass is 2.26 Å [1] what is the evidence for the nearest neighborhood of Ge and S atoms. The structural groups are arranged in clusters with intercluster distances ~7.6 Å.

The short-range order parameters for glassy Sb<sub>2</sub>S<sub>3</sub> were determined from the RADC maxima (Fig. 1b), calculated from the experimental curve of scattered X-ray intensity I(s) Fig. 1a). The presence of the main maximum is determined by strong chemical interactions of the atoms. Its splitting into two maxima (2.10 and 2.50 Å<sup>-1</sup>) is related to the existence of two types of atomic distributions. At low *s* values (~1.20 Å<sup>-1</sup>) the *I*(*s*) function exhibits a diffraction maximum caused by the presence of large scattering complexes [2]. For GeS<sub>2</sub> glass this maximum is located at  $s \approx 1.02$  Å<sup>-1</sup> [3].



Fig. 1. Scattered X-ray intensity curve (a) and radial atomic distribution curve (b) for glassy Sb<sub>2</sub>S<sub>3</sub>.

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Sb<sub>2</sub>S<sub>3</sub> glass RADC is characterized by three distinct maxima at  $R_1 = 2.50$  Å,  $R_2 = 3.86$  Å,  $R_3 = 4.64$  Å. The  $R_1$  value, close to the sum of Sb and S atoms covalent radii, is the evidence for the nearest neighborhood. This enables us to assume the Sb<sub>2</sub>S<sub>3</sub> glass matrix to be constructed mainly of SbS<sub>3</sub> trigonal pyramids, linked together by two-fold coordinated sulphur atoms. However, the calculated values of coordination numbers for Sb<sub>2</sub>S<sub>3</sub> glass ( $z_S(Sb) = 4.0$ ;  $z_{Sb}(S) = 2.4$ ) are higher than for the crystal (3 and 2) what is probably related to the presence of a great number of structural groups with homopolar bonds (S-S and Sb-Sb) in its structure. The obtained results show a good correlation with the vibrational spectroscopy data. The IR spectrum of the Sb<sub>2</sub>S<sub>3</sub> glass consists of an intense broad absorption band with a maximum at ~ 290 cm<sup>-1</sup> and a weak absorption band in the range of ~ 190 cm<sup>-1</sup> (Fig. 2, curve 7). The intense absorption band results from the vibrations of covalently bonded atomic pairs in SbS<sub>3</sub> trigonal pyramids while the weak band at 190 cm<sup>-1</sup> is due to the presence of structural groups with homopolar S-S and Sb-Sb bands. This assumption is also confirmed by the results of Raman scattering studies of Sb-S glass system [4].



Fig. 2. IR transmission spectra of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses: x = (1), 0.9 (2), 0.8 (3), 0.50 (4), 0.30 (5), 0.10(6), 0 (7).

The structure of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  (0 <*x*<1) glasses somewhat differs from the structure of GeS<sub>2</sub> and Sb<sub>2</sub>S<sub>3</sub> glasses. Far-infrared transmission spectra for this glass system are shown in Fig. 2. The main changes in the spectra with the increase of Sb<sub>2</sub>S<sub>3</sub> content occur in the high - frequency range. Here for glassy GeS<sub>2</sub> three overlapping bands in the frequency range from 320 to 450 cm<sup>-1</sup> are observed (Fig. 2, curve 1). Already at the first admixtures of Sb<sub>2</sub>S<sub>3</sub> to GeS<sub>2</sub> the bands in the range of 450 cm<sup>-1</sup> are revealed. With the increase of Sb<sub>2</sub>S<sub>3</sub> content these bands grow in intensity and gradually shifts to lower frequencies. For glassy Sb<sub>2</sub>S<sub>3</sub> the band maximum is located at 290 cm<sup>-1</sup>. Meanwhile the band at 375 cm<sup>-1</sup> vibrations of atomic pairs in tetrahedral structural groups GeS<sub>4</sub> shifts to higher frequencies. The predominantly two-mode character of vibrational spectra transformation was also observed at Raman studies of GeS<sub>2</sub>-Sb<sub>2</sub>S<sub>3</sub> glass system [5]. It should be noted that in the IR spectra of some of the (GeS<sub>2</sub>)<sub>x</sub>(Sb<sub>2</sub>S<sub>3</sub>)<sub>1-x</sub> glasses with a considerable content of germanium disulphide the weak bands in the range of ~240 cm<sup>-1</sup> are observed (Fig. 2, curves 2 - 4). They can be related to the presence of a small amount of S<sub>3</sub>Ge-GeS<sub>3</sub> structural groups in the glass matrix.

Thus, the analysis of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glass system vibrational spectra enables one to assume the formation of a single glass network built by  $GeS_4$  and  $SbS_3$  structural groups which are linked together by sulphur atoms.

The investigation of the absorption coefficient spectra for  $GeS_2-Sb_2S_3$  glasses has shown that in the fundamental absorption range they are well described by the Urbach rule  $\alpha = \alpha_0 exp\Gamma$  (hv - hv<sub>0</sub>). The  $\Gamma$  parameter varies with the glass composition from 13 eV<sup>-1</sup> for GeS<sub>2</sub> to 18 eV<sup>-1</sup> for Sb<sub>2</sub>S<sub>3</sub>, this being the evidence of the difference in the short-range order for these compounds. The optical gap value  $E_g$ , determined from the spectral plot of  $\alpha$ , varies depending on the composition of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses from 1.65 eV for  $Sb_2S_3$  to 3.06 eV for  $GeS_2$ . The values of  $E_g$  for the  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses series are listed in Table 1. As shown in [6], the compositional dependence of  $E_g$  for glasses of this system is non-linear. This is related to the fact that already small admixtures of  $Sb_2S_3$  (GeS<sub>2</sub>) result in strong distortion of  $GeS_4$  tetrahedra (SbS<sub>3</sub> pyramids) and shortening of tetrahedral (trigonal) chains being the main structural pattern of  $GeS_2$  (Sb<sub>2</sub>S<sub>3</sub>) glass. Strong fluctuations of the bond lengths and angles resulting from this distortion, lead to additional broadening of the density-of-states tails at the allowed band edges and, as a result, to the deviation of the  $E_g(x)$  dependence from linearity.

Parameters	x, mol.%				
	1.00	0.96	0.80	0.70	0.50
E <sub>g</sub> , eV	3.06	3.00	2.78	2.62	2.37
$\epsilon(10^5 \text{ Hz})$	5.3	6.8	8.0	8.6	11.7
$\rho \cdot 10^{-3}, \text{ kg/m}^3$	2.84	2.90	3.14	3.30	3.60
V <sub>l</sub> , m/s	2778±10	2698±10	2670±10	2660±10	2622±10
V <sub>t</sub> , m/s	1784±20	1641±20	1589±20	1484±20	1438±20
μ	0.149	0.206	0.226	0.283	0.285
$G \cdot 10^{-9}$ , Pa	9.04	7.81	7.93	7.07	7.44
$\mathrm{E}\cdot10^{-9}$ , Pa	20.7	18.8	19.4	18.1	19.1
$a \cdot 10^{11}, Pa^{-1}$	10.2	9.38	8.47	7.19	6.75
K · 10 <sup>-9</sup> , Pa	9.8	10.6	11.8	13.9	14.8
$M_2 \cdot 10^{15}$ , s <sup>3</sup> /kg	110	134	314	325	340
$\gamma(100 \text{ MHz}), \text{ dB/cm}$	1.20	1.33	1.33	1.40	1.41

The strong deformations of the structural groups, forming the glass matrix, at the admixture of the other component result not only in the changes in optical properties, but also to the variation of other parameters. The measurements have shown that the density  $\rho$  of the  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses varies from  $2.84 \times 10^3$  kg/m<sup>3</sup> to  $4.14 \times 10^3$  kg/m<sup>3</sup>, and the dielectric permeability  $\varepsilon$  - from 5.3 to 28.4. Especially considerable variation of the glass parameters occurs in the range of compositions, close to the binary compounds Sb<sub>2</sub>S<sub>3</sub> and GeS<sub>2</sub>. In the compositional interval 0.05 <*x*< 0.10 a knee at the  $\rho(x)$  plot is observed, as well as a maximum at the compositional dependence of  $\varepsilon$ . In the vicinity of GeS<sub>2</sub> glass the changes in  $\rho$  and  $\varepsilon$  with composition are smaller (Table 1).

The variation of structure and composition is also revealed in the photoelastic properties of  $(GeS_2)_x(Sb_2S_3)_{1-x}$  glasses. The values of longitudinal  $(v_1)$  and transverse  $(v_t)$  ultrasonic wave propagation rates, measured by coherent optical radiation diffraction on ultrasonic waves in the sample [7], are listed in Table 1 as well as the acoustooptical figure of merit  $M_2$ , measured by Dixon-Cohen method.

The photoelastic parameters of the glasses, calculated from the expressions [6]

$$\mathbf{K} = \frac{\rho \left( 3v_1^2 - 4v_t^2 \right)}{3}; \quad \mathbf{E} = \rho v_t^2 \frac{\left( 3v_1^2 - 4v_t^2 \right)}{\left(v_1^2 - v_t^2 \right)}; \tag{1}$$

$$G = \rho v_t^2; \quad \mu = \frac{\left(v_1^2 - 2v_t^2\right)}{2\left(v_1^2 - v_t^2\right)}; \quad \aleph = \frac{1}{K}.$$
 (2)

are also given. Here K, E, G, are bulk, Young's and shear moduli,  $\mu$  - Poisson coefficient,  $\aleph$  - adiabatic compressibility.

It is seen from Table 1 that glassy germanium disulphide possesses the highest values of Young's and shear moduli. This is, perhaps, related to the highest linking degree of GeS<sub>2</sub> glass among the chalcogenide glassy semiconductors. Under the ultrasonic wave the structural groups in the glass are distorted mostly along the pressure direction. This is also confirmed by the compositional dependences of the Poisson coefficient. Introduction of Sb<sub>2</sub>S<sub>3</sub> into GeS<sub>2</sub> and increase of its content in the glass composition results in the decrease of *E* and *G* moduli, providing the evidence for the decrease of linking in the (GeS<sub>2</sub>)<sub>x</sub>(Sb<sub>2</sub>S<sub>3</sub>)<sub>1-x</sub> glass network.

The acoustooptical figure of merit  $M_2$  increases with Sb<sub>2</sub>S<sub>3</sub> content in the glass and at x = 0.50 reaches the values, characteristic for the best chalcogenide glass compositions, applied as acoustic lines in acoustooptical devices [9]. The damping,  $\gamma$ , of ultrasonic waves of the investigated glasses is relatively low (Table 1).

Thus, based on the performed studies, one may conclude that due to the transparency in the visible spectral range, high acoustooptical figure of merit, small damping of ultrasonic waves and significant mechanical strength, Ge-Sb-S glasses can be used as active elements in acoustooptical devices.

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